

Qualitative Analysis (MCQ)

1. Four pairs of solutions are mixed.

Which pair of solutions forms a white precipitate?

- A $\text{NH}_4\text{Cl}(\text{aq})$ and $\text{NaOH}(\text{aq})$
- B $\text{KBr}(\text{aq})$ and $\text{AgNO}_3(\text{aq})$
- C $\text{FeCl}_3(\text{aq})$ and $\text{NH}_3(\text{aq})$
- D $\text{Cr}_2(\text{SO}_4)_3(\text{aq})$ and $\text{BaCl}_2(\text{aq})$

Your answer

[1]

END OF QUESTION PAPER

Mark scheme – Qualitative Analysis (MCQ)

Question			Answer/Indicative content	Marks	Guidance
1			D	1	
			Total	1	